

## Novel Insertion of the Dioxophosphorane Moiety ('Metaphosphonate') into an Aliphatic C–H Bond *via* Flash Vacuum Pyrolysis of a Cyclic Phosphonite and by Oxidation of a Diphosphene

J. I. G. Cadogan,<sup>a</sup> Alan H. Cowley,<sup>b</sup> Ian Gosney,<sup>c</sup> Marek Pakulski,<sup>b</sup> and Salih Yaslak<sup>c</sup>

<sup>a</sup> BP Research Centre, Chertsey Road, Sunbury-on-Thames, Middlesex TW16 7LN, England

<sup>b</sup> Department of Chemistry, University of Texas, Austin, Texas 78712, U.S.A.

<sup>c</sup> Department of Chemistry, University of Edinburgh, West Mains Road, Edinburgh EH9 3JJ, Scotland

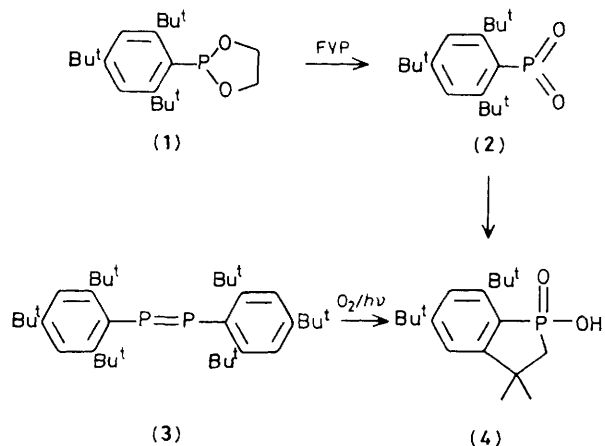
Flash vacuum pyrolysis of 2-(2,4,6-tri-*t*-butylphenyl)-1,3,2-dioxaphospholane (**1**) gave 5,7-di-*t*-butyl-3,3-dimethyl-2,3-dihydro-1-hydroxy-1 $\lambda^5$ -benzophosphol-1-one (**4**), 100%, also produced by photo-oxidation of bis(2,4,6-tri-*t*-butylphenyl)diphosphene (**3**) *via* insertion of the dioxophosphorane ( $-\text{PO}_2$ ) moiety into a neighbouring C–H bond.

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There is considerable current interest in unusually bonded phosphorus compounds, particularly in dioxophosphoranes

( $\text{RPO}_2$ )<sup>1</sup> and diphosphenes ( $\text{RP}=\text{PR}$ ).<sup>2</sup> We now report novel experiments which link both species.

Thus, following the recent generation of aryldioxo-phosphoranes by flash vacuum pyrolysis (F.V.P.) of cyclic arylphosphonites,<sup>3</sup> F.V.P. at 700 °C and 0.01 mmHg of 2-(2,4,6-tri-*t*-butylphenyl)-1,3,2-dioxaphospholane (1),† prepared from 2-chloro-1,3,2-dioxaphospholane and 2,4,6-tri-*t*-butylphenyl-lithium by a standard route, gave the cyclic phosphinic acid (4) (100%, m.p. 320–323 °C) which was



Scheme 1

† All new compounds had correct analytical and spectroscopic characteristics.

identical with material prepared by a different route.<sup>4</sup> This is the first reported example of the formal insertion of the highly electrophilic  $PO_2$  moiety into an alkyl group and is analogous to insertion of the two-co-ordinated phosphonium ion, produced by protonation of a diaryldiphosphene,<sup>5</sup> also into a neighbouring *t*-butyl group.

The connection with the diphosphene system is further highlighted by the novel photo-oxidation of bis(2,4,6-tri-*t*-butylphenyl)diphosphene (3)<sup>6</sup> which gave the same cyclic phosphinic acid (4) albeit in 10% yield. The reactions shown in Scheme 1 are suggested to occur with the intriguing possibility that the dioxophosphorane species (2) is also involved in the photo-conversion of (3) into (4).

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